

1. A composite ion selective electrode comprising

- in which the calcium ion selective electrode member comprises an electro-insulating support, a pair of electrode units each of which comprises a silver layer and a silver halide layer, and which are electrically separated from each other, an electrolyte layer, and a calcium ion selective membrane, and

- (2) an electro-insulating member having two openings in which one opening is provided for introducing a sample liquid into the composite electrode and another opening is provided for introducing a reference liquid into the composite electrode;

- (4) a bridge member which is provided on the electro-insulating member to bridge the two openings of the electro-insulating member so as to electrically connect the introduced sample liquid and the introduced reference liquid;

which is characterized in that the calcium ion selective membrane has a thickness of 5 to 30 μm , and the hydrogen ion selective membrane contains tri-n-dodecylamine and trisethylhexyl trimellitate.

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2. The composite ion selective electrode of claim 1, wherein the calcium ion selective membrane contains calcium di[4-(1,1,1,3-tetramethylbutyl)phenyl] phosphate.

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3. The composite ion selective electrode of claim 2, wherein the calcium ion selective membrane comprises a vinyl chloride-vinyl acetate copolymer, dioctylphenyl phosphonate, and calcium di[4-(1,1,1,3-tetramethylbutyl)-phenyl] phosphate.

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4. The composite ion selective electrode of claim 1, wherein the thickness of calcium ion selective membrane is in the range of 5 to 20 μm .

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5. The composite ion selective electrode of claim 4, wherein the thickness of calcium ion selective membrane is in the range of 10 to 18 μm .

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6. The composite ion selective electrode of claim 1, wherein the hydrogen ion selective membrane has a thickness of 5 to 30 μm .

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7. The composite ion selective electrode of claim 1, wherein the electrolyte layer comprises sodium chloride.

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8. A method for determining a standardized calcium ion concentration in a sample blood, which comprises the steps of:
spotting a sample blood and a reference liquid onto openings of the electro-insulating member of the compos-

ite ion selective electrode of claim 1, respectively;
measuring potentiometrically a calcium ion concentration and a hydrogen ion concentration in the sample blood; and

5 incorporating the measured calcium ion concentration and the measured hydrogen ion concentration into the following equation to obtain a value of Log (standardized iCa):

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$$\text{Log (standardized iCa)} = \text{Log (iCa at pH)} - 0.22 \times (7.4 - \text{pH})$$

in which iCa means a calcium ion concentration and pH means a hydrogen ion concentration.

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9. A calcium ion selective electrode comprising

(1) a calcium ion selective electrode member which comprises an electro-insulating support, a pair of electrode units each of which comprises a silver layer and a silver halide layer, and which are electrically separated from each other, an electrolyte layer, and a calcium ion selective membrane;

(2) an electro-insulating member having two openings which is provided on the calcium ion selective electrode member and in which one opening is provided for introducing a sample liquid into the composite electrode and another opening is provided for introducing a reference liquid into the composite electrode; and

(3) a bridge member which is provided on the electro-insulating member to bridge the two openings of the electro-insulating member so as to electrically connect the introduced sample liquid and the introduced reference liquid;

which is characterized in that the calcium ion selective membrane has a thickness of 5 to 30 μm .

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10. The calcium ion selective electrode of claim 9, wherein the calcium ion selective membrane contains calcium di[4-(1,1,1,3-tetramethylbutyl)phenyl] phosphate.

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11. The calcium ion selective electrode of claim 10, wherein the calcium ion selective membrane comprises a vinyl chloride-vinyl acetate copolymer, dioctylphenyl phosphonate, and calcium di[4-(1,1,1,3-tetramethylbutyl)-phenyl] phosphate.

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12. The calcium ion selective electrode of claim 10, wherein the thickness of calcium ion selective membrane is in the range of 5 to 20 μm .

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13. The calcium ion selective electrode of claim 12, wherein the thickness of calcium ion selective membrane is in the range of 10 to 18 μm .

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14. The calcium ion selective electrode of claim 10, wherein the electrolyte layer comprises sodium chloride.

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15. A hydrogen ion selective electrode comprising (1) a hydrogen ion selective electrode member which comprises an electro-insulating support, a pair of electrode units each of which comprises a silver layer and a silver halide layer, and which are electrically separated from each other, an electrolyte layer, and a hydrogen ion selective membrane;

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(2) an electro-insulating member having two openings which is provided on the hydrogen ion selective electrode member and in which one opening is provided for introducing a sample liquid into the composite electrode and another opening is provided for introducing a reference liquid into the composite electrode; and

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(3) a bridge member which is provided on the electro-insulating member to bridge the two openings of the electro-insulating member so as to electrically connect the introduced sample liquid and the introduced reference liquid;

which is characterized in that the hydrogen ion selective membrane contains tri-n-dodecylamine and tri-ethylhexyl trimellitate.

16. The hydrogen ion selective electrode of claim 15, wherein the hydrogen ion selective membrane comprises tri-n-dodecylamine, trisethylhexyl trimellitate, potassium tetrakis(p-chlorophenylborate) and a vinyl chloride-vinyl acetate copolymer.

17. The hydrogen ion selective electrode of claim 15, wherein the hydrogen ion selective membrane has a thickness of 5 to 30 μm .

18. The composite ion selective electrode of claim 15, wherein the electrolyte layer comprises sodium chloride.